

1. Volume $V=100000 \mu^3+10000\mu^3 \approx 10^5 \mu^3$. Number of protein molecules is $N=500 \text{ molecules/minute} \times 60 \text{ minutes/hour} \times 24 \text{ hours/day} \times 7 \text{ days/week} / N_a$, N_a - Avogadro number
Concentration $C=N/V=8.4 \times 10^{-8} \text{ M}$
2. Dynamic equilibrium: $dN_{\text{synthesized molecules}}/dt=dN_{\text{transported to axon}}/dt=1000 \text{ molecules/minute}=K$
 $dN_{\text{transported to axon}}/dt=v \cdot A \cdot C$, v -rate of transport (5 mm/day), A -cross-sectional area, C stable concentration
 $C=K/(v \cdot A)=2.4 \times 10^{-7} \text{ M}$
3.
 - a. $2Dt=\langle \Delta X^2 \rangle$, $t=(0.05 \text{ cm}^2)^2/2D=6.25 \times 10^4 \text{ s}$
 - b. Presumably the diffusion doesn't have directional preference, so half of synthesized molecules will be in dendrite, and half will be in axon.
In dendrite: $C=(N/2)/(N_a \cdot V_{\text{dendrite}}) \approx 21 \mu\text{M}$, volume $V_{\text{dendrite}}=4 \mu^2 \cdot 1 \text{ mm}$
In axon: the first 1 mm ($\approx 2 \cdot \langle \Delta X^2 \rangle$) contains ~95% of molecules diffused in axon (two standard deviations of normal distribution), so average concentration $C=0.95 \cdot (N/2)/(N_a \cdot V_{\text{axon}}) \approx 79 \mu\text{M}$, volume $V_{\text{axon}}=1 \mu^2 \cdot 1 \text{ mm}$
4. $\text{pH}_{\text{medium}}=7.4$, so $[\text{H}^+]=10^{-7.4} \text{ M}=40 \text{ nM}=C_c$
Surface is negatively charged to $V=-50 \text{ mV}$, and negative charges are distributed by Boltzmann distribution: surface concentration of positive ions (protons)
 $C_s=C_c \cdot \exp[zeV/kT]$, and surface $\text{pH}=-\lg C_s=6.5$
5. The surface area of sphere is proportional (" ∞ ") to the r^2 . We have ~25% of radius change ($[7.5 \mu-6 \mu]/6 \mu$), so we have $(1.25)^2 \approx 1.56$ of initial membrane area (~56% stretching). But even 4% stretching leads to lysis, so lymphocytes can swell due to smoothing of folds in membrane.
6. At the beginning we have K^+ -defined membrane potential:
 $V_K=RT/(zF) \ln[3 \text{ mM}/140 \text{ mM}] \approx -90 \text{ mV}$, where is $z=1$. At the end we can calculate voltage difference again by Nernst equation, but for Ca^{2+} :
 $V_{\text{Ca}}=RT/(zF) \ln[0.5 \text{ mM}/2 \mu\text{M}] \approx +66 \text{ mV}$, where is $z=2$. The change in potential will be $\Delta V=V_K-V_{\text{Ca}} \approx -156 \text{ mV}$
7. Donnan equilibrium (see also Lecture#11): $C_{\text{K inside}}/C_{\text{K outside}}=C_{\text{Cl outside}}/C_{\text{Cl inside}}$, Also we know that $C_{\text{K outside}}=1 \text{ mM}$ $C_{\text{Cl outside}}=110 \text{ mM}$. Another equation – electroneutrality inside the cell (assuming that there are only K & Cl):
 $C_{\text{K inside}}=C_{\text{Cl inside}}$. Also x – change in concentrations inside: $C_{\text{K inside}}=140+x$, and $C_{\text{Cl inside}}=4+x$. By substituting one equation to another: $x^2+144x+450=0$, the meaningful solution of this square equation (which will keep concentrations positive) $x=-3.2 \text{ mM}$, so $C_{\text{Cl inside}}=4+x=0.8 \text{ mM}$, and $C_{\text{K inside}} \approx 136.8 \text{ mM}$.
Potential difference by Nernst equation (for example for potassium):
 $V=56 \text{ mV} \cdot \lg[1/136.8] \approx -118 \text{ mV}$.

- 8.
- a. Bending energy for the membrane with constant curvature
 $E = (B/2) * (1/R_t)^2 * A$. Area of cylinder $A = \pi * d * l \approx 5 \mu^2$
 $E = 7.8 * 10^{-17} \text{ N*m}$, $KT \approx 4 * 10^{-21} \text{ N*m}$ ($T = 300\text{K}$)
 So $E \approx KT * 10^4$
 - b. Affinity of enhancer (E) to DNA gives us the change in free energy: $\Delta G_1 = RT * \ln[K_{E-DNA}] \approx 5.5 \text{ Kcal/mole}$. For binding E to transcription factor (TF) we know change in entropy and enthalpy of binding, so we also can calculate change in free energy: $\Delta G_2 = \Delta H - T * \Delta S = -4 - 1 = -5 \text{ Kcal/mole}$. For full reaction: $\Delta G = \Delta G_1 + \Delta G_2 = -10.5 \text{ Kcal/mole}$, and equilibrium constant: $K_A = \exp[-\Delta G / (RT)] \approx 4 * 10^7 \text{ 1/M}$, and $K_D = 2.5 * 10^{-8} \text{ M}$; We also know that we need enhancer concentration equal to dissociation constant for 50% activation of transcription. By knowing the volume of nucleus we can calculate the actual number of molecules: $N = K_D * N_A * V = \approx 1700 \text{ molecules}$
9. $D = kT / (6\pi\eta r) = 2.2 * 10^{-7} \text{ cm}^2/\text{s}$ (You also can calculate it by using the $D \propto 1/r$ proportionality from this formula (see comments for (5) of the provided to exam equations): $D_{1\mu \text{ radius sphere}} = 4.4 * 10^{-9} \text{ cm}^2/\text{s}$, so $D_{0.05\mu \text{ radius virus particle}} = 2.2 * 10^{-8} \text{ cm}^2/\text{s}$)
10. The first mechanism – linear movement of scanning protein complex with the rate $v = 30 \text{ b.p./s} \approx 6 * 10^{-4} \text{ cm/s}$ (3 \AA/b.p.), so scanned distance will be $l = t * v$
 For diffusion model scanned distance defined by $d = \Delta X / 2$, and $\langle \Delta X^2 \rangle = 2D * t$.
 For $t = 1 \text{ min} = 60 \text{ s}$: $l \approx 5.4 * 10^{-2} \text{ cm} \approx 1800 \text{ b.p.}$, and $d = (1/2) * (2DT)^{1/2} \approx 1.7 \mu\text{m} \approx 5800 \text{ b.p.}$
 For $t = 1 \text{ h} = 3600 \text{ s}$: $l \approx 3 \text{ cm} = 108 \text{ Kb.p.}$, $d \approx 13.4 \mu\text{m} = 45 \text{ Kb.p.}$
11. $L = 5000 \text{ bp}$, rate of synthesis is $v = 50 \text{ bp/s}$, so time of synthesis is $T = L/v = 100 \text{ s}$.
 Time between moment of transcription initiations $t = 300 \text{ bp}/v = 6 \text{ s}$. 100^{th} copy of RNA will be synthesized in $99 * 6 = 594^{\text{th}}$ second after beginning of synthesis of the first RNA molecule, and will finish synthesis in next $T = 100 \text{ s}$, so cycle for synthesis of 10 molecules is $t + T \approx 700 \text{ s}$. And the gene should be activated $700 \text{ s}/30 \text{ min} \approx 39\%$ of time.
12. Reynold's number: $R = vL\rho/\eta$, L - here is diameter, v - speed, ρ - density of object (particles), η - viscosity of media (we can use viscosity of water here – brain slices are usually covered by thin layer of some liquid). So $v \approx 1.7 \text{ m/s}$