

JACS Year in Review: 1978

Laura Schacherer

Statistics for *JACS* 1978

Total Number of Papers: **1,947**

Most Cited Paper: p.3686 (**cited 1396x!** Ammeter et al
“**Counterintuitive orbital mixing in semi-empirical and ab initio molecular-orbital calculations.**”)

Top 25 Authors:

24: Michl J

16: Paquette LA

12: Cotton FA, Dewar MJS

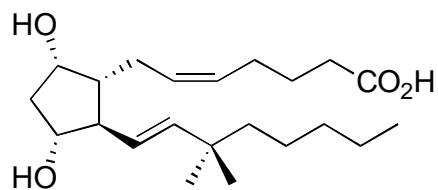
10: Arnett EM, Chisholm MH, Houk KN, Roberts JD, Seff K, Turro NJ

9: Epiotis ND, Hoffmann R, Ibers JA

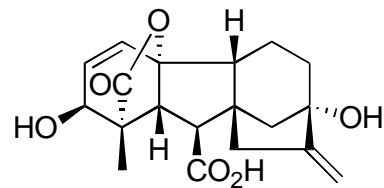
8: Schrock RR, Streitwieser A, Trost B,

7: Bard AJ, Beuchamp JL, Brauman JI, Corey EJ, Hayakawa Y, Noyori R, Olah GA, Radom L, Rzepa HS

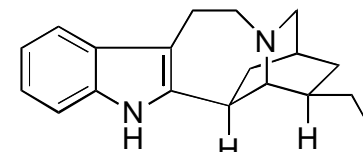
Some Synthetic Achievements in 1978:



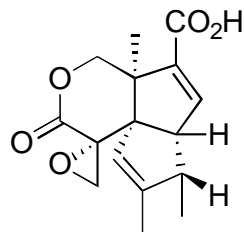
Prostaglandin F₂α: Stork, 8272



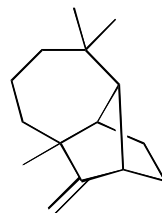
Gibberellic Acid: Corey, 8034



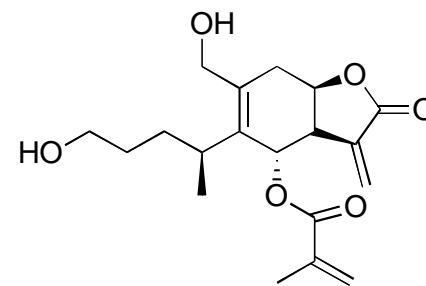
Ibogamine: Trost, 3930



dl-Pentalenolactone: Danishefsky, 6536

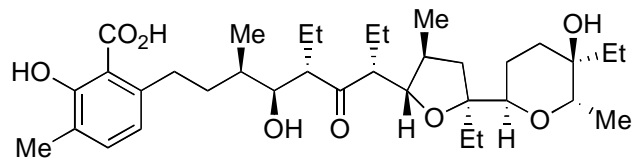


Longifolene: Oppolzer, 2583

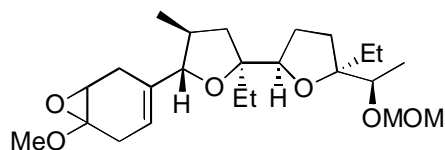


Eriolanin: Grieco, 1616

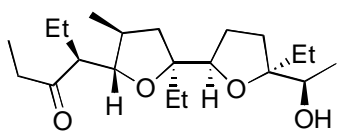
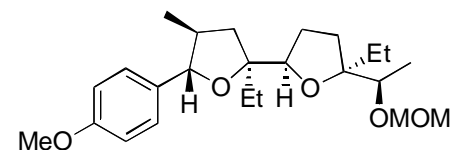
Lasoloid A: Kishi, 2933



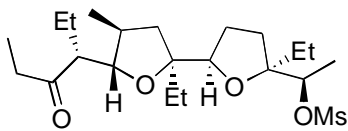
HIO_4 , aq. dioxane
 LAH
 TsCl, pyr.
 LAH
 B_2H_6 , THF
 Jones Ox.
 TrBF_4 , DCM



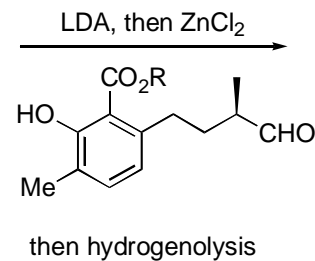
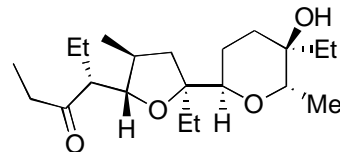
Li , EtOH, NH_3
 MCPBA, NaHCO_3



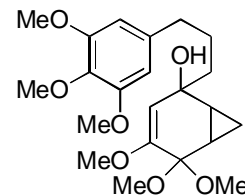
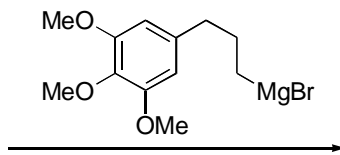
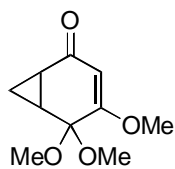
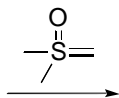
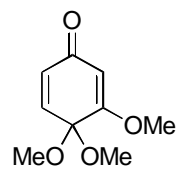
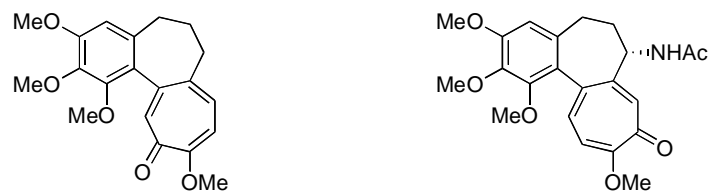
NaOH
 MsCl , pyr.



Ag_2CO_3 ,
 aq. acetone

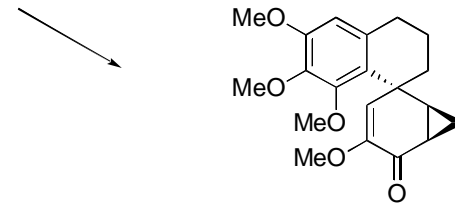
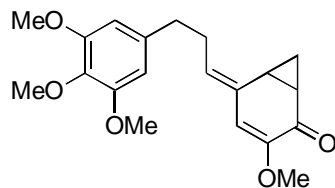
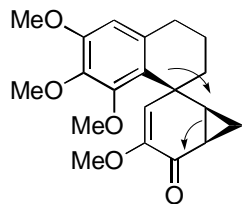


Colchicine: Evans, 4593

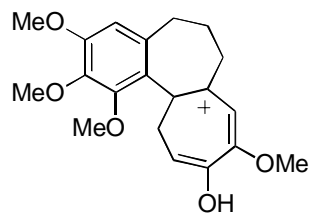


Pdt

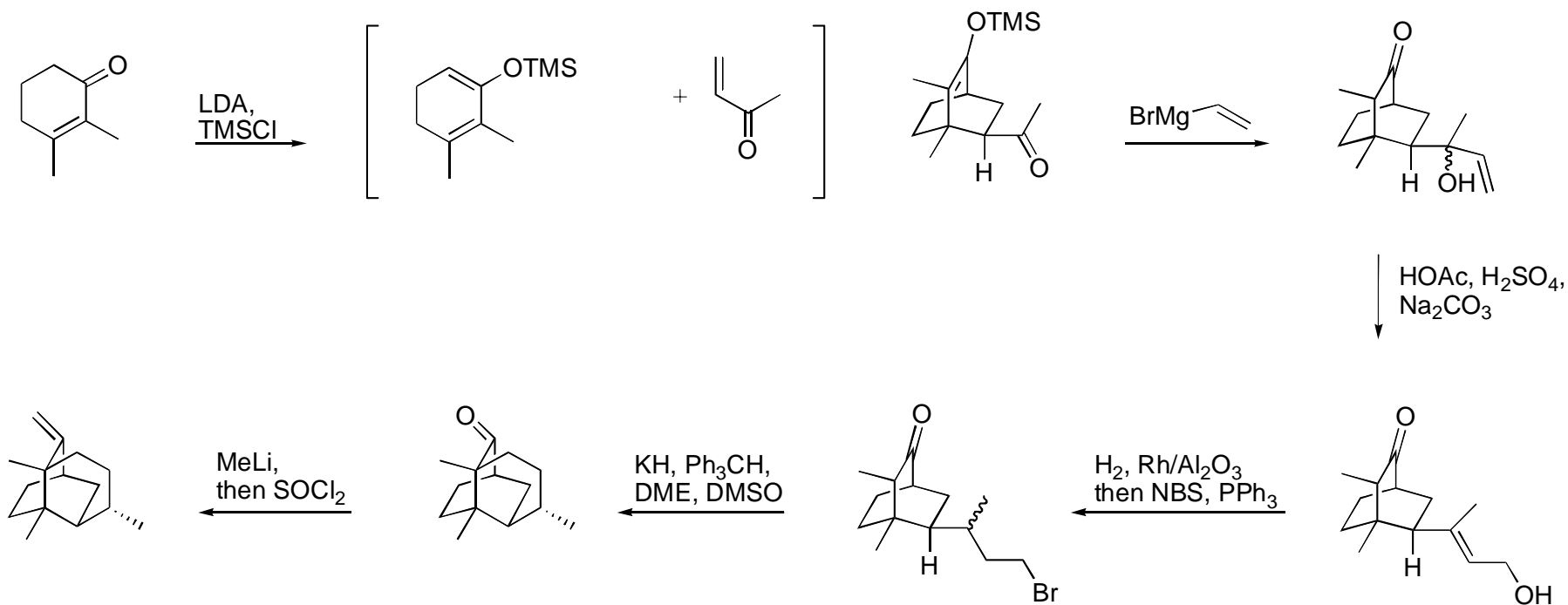
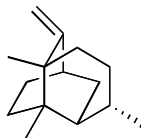
Via:



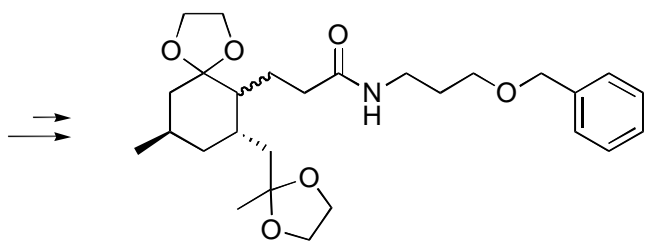
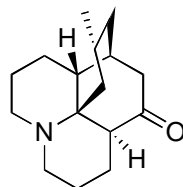
X



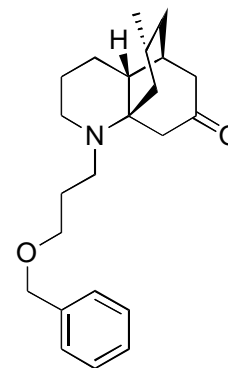
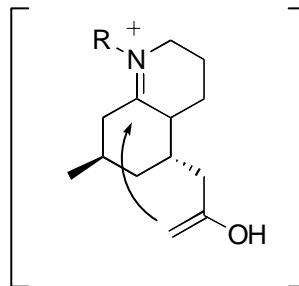
Seychellene: Jung, 5207



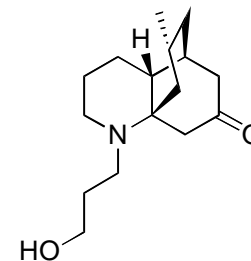
Lycopodine: Heathcock



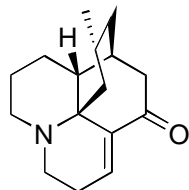
LAH,
then HCl, 2 wks!



H₂, Pd, EtOH, HCl



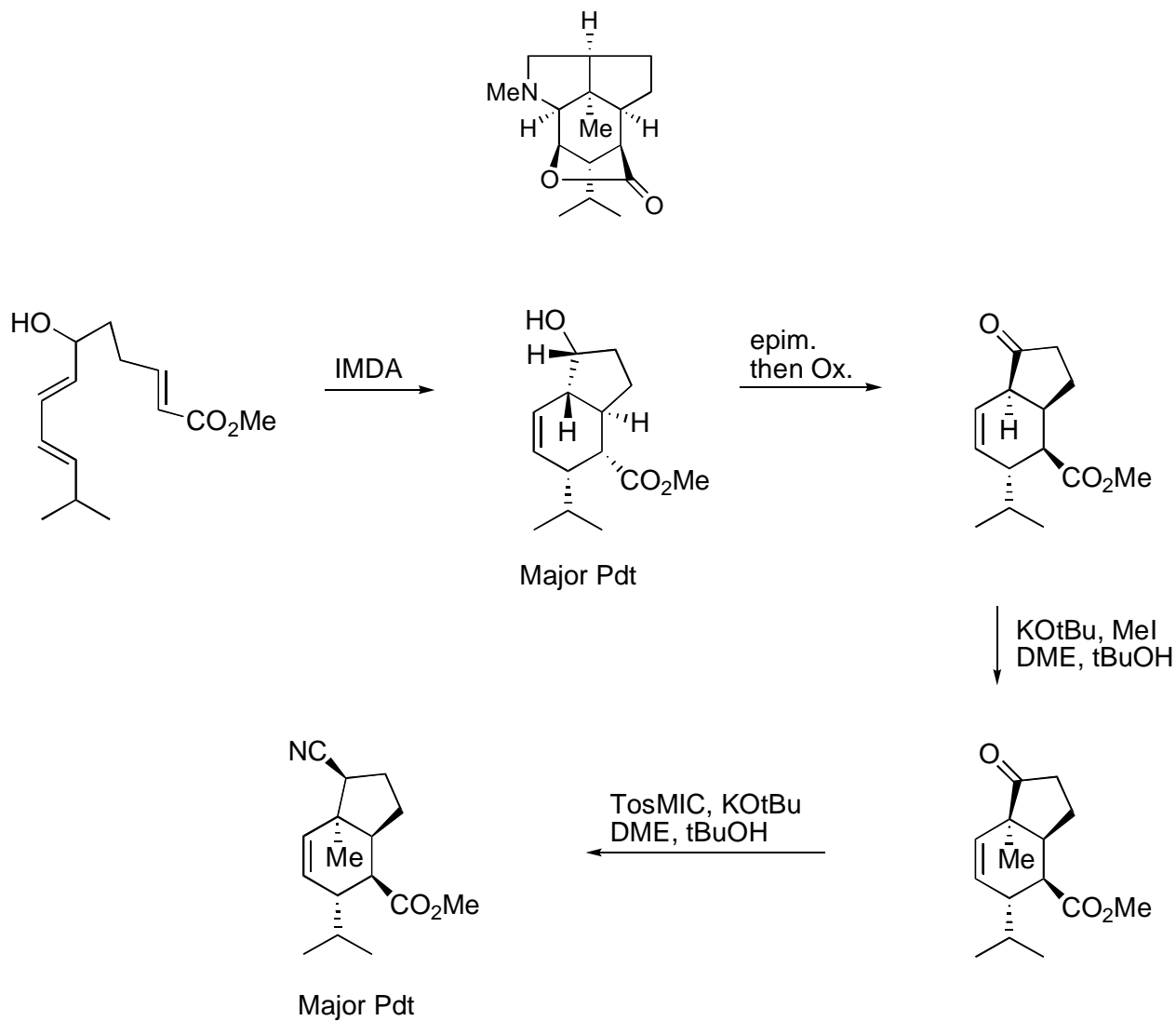
Oppenauer Oxidation:
Benzophenone, benzene, Δ
(then intramolecular aldol
and dehydration)



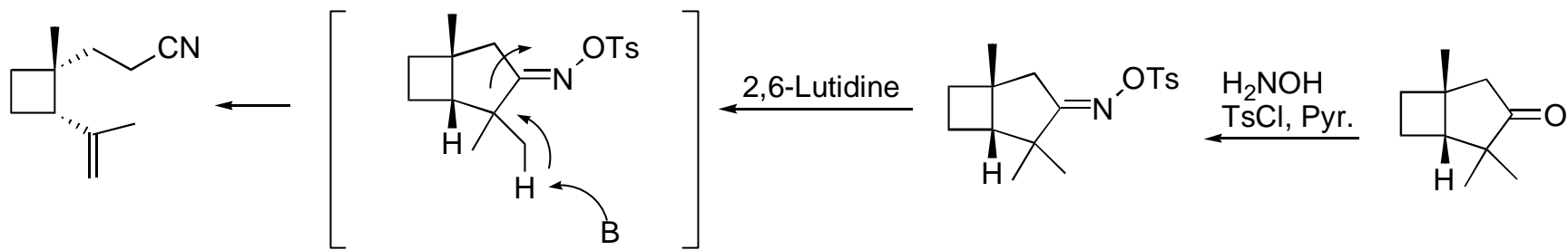
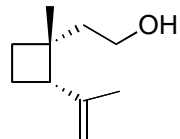
H₂, Pt, EtOH

→

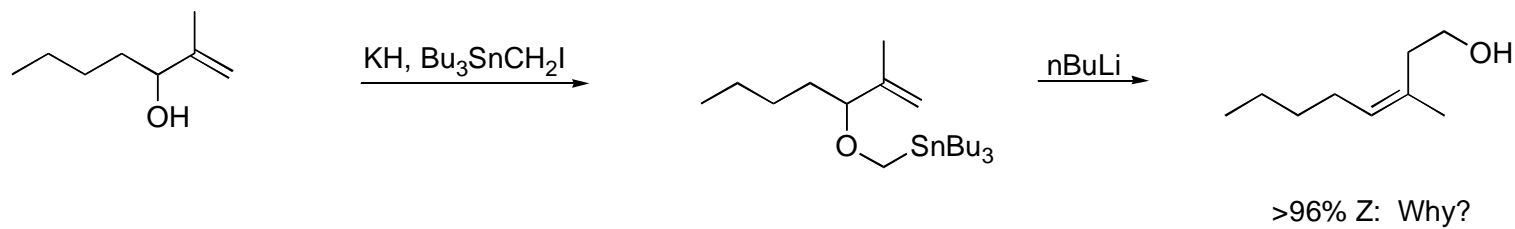
Dendrobine: Rousch



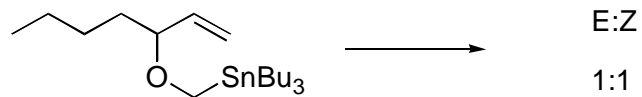
Grandisol : Golob



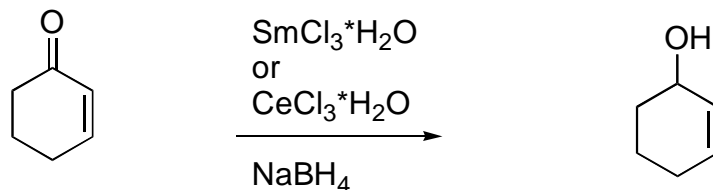
Methodology: Still, 1927



Further evidence:



Methodology: Luche Reduction, 2226



Mechanistic Information:

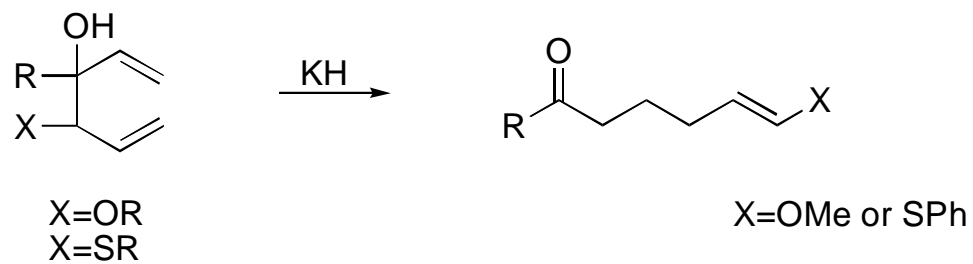
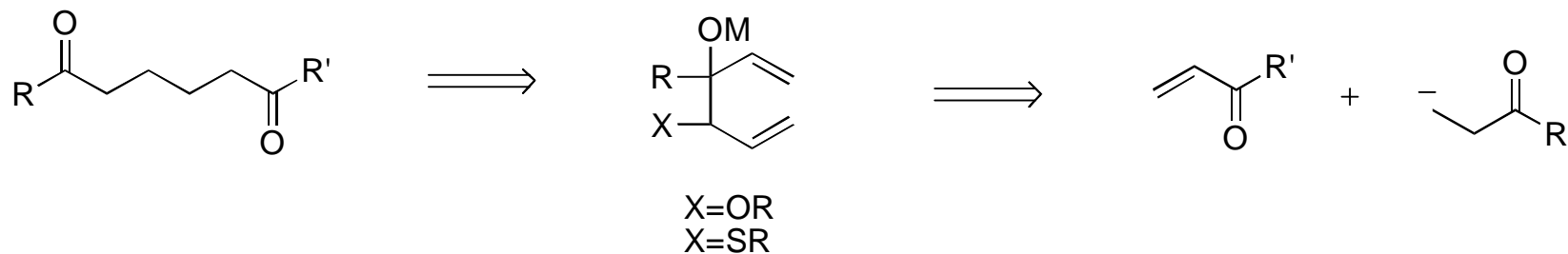
Catalytic in Lanthanide

Could be Lanthanide hydride formed *in situ*

or

Ketone/Lanthanide Complexation

Methodology: Evans, 2242



Methodology: Tebbe, 3611

