## SOME IMPORTANT INFORMATION

Chemistry C2407X	TTh
George Flynn	1:10 p.m.

All information contained in this handout can be found at the C2407 course website: http://www.columbia.edu/itc/chemistry/chem-c2407/

PLEASE NOTE: FOR STUDENTS IN FU-SEAS: THE LAST DAY TO DROP THIS COURSE IS NOVEMBER 13, 2003 FOR STUDENTS IN COLUMBIA COLLEGE: THE LAST DAY TO DROP THIS COURSE IS OCTOBER 7, 2003

 (1) <u>Exams:</u> Three (3) "hour" exams (75 minutes, 1 class period) will be given on the following days: Thursday, September 25, 2003 (covers material through lecture of 9/18/03) Thursday, October 23, 2003 (covers material through lecture of 10/16/03) Tuesday, November 25, 2003 (covers material through lecture of 11/18/03)

Note that there is generally a week between the end of the lecture material and each exam. I will actually announce in class the cut off for lecture material for each exam. The above is, however, approximately correct. The timing of the exam given during Thanksgiving week is a compromise among bad choices driven by a change (by one week) in the scheduling of the Thanksgiving Holiday by the U.S. Congress.

- (2) <u>Recitations:</u> There will be a recitation, Monday, 3-5 or 6-8 p.m., or Tuesday, 3-5 p.m., or Wednesday, 4-6 p.m., or Friday, 10 a.m.-12 noon or 2-4 p.m. These will be run by our graduate teaching assistants, Jennifer Inghrim (854-4964) (jai2002@columbia.edu), and Sean Moran (854-8468) (sdm2007@columbia.edu). Use the phone registration system to sign up for these recitations. The course number for the recitations is C2409. Seven of these recitations are required and 3 will be used to review for the class period exams described above. The other 4 weeks of the term, the graduate teaching assistants will have expanded office hours during recitation times to answer questions you may have about course material.
- (3) <u>Grades:</u> The final exam will be counted 36%, each "hour" exam 18%, and recitation 10% in determining your final grade for the course.

- (4) <u>Absence from Exams</u>: <u>There are no make-up exams</u> for the "hour" (class period) exams. If you miss one of these exams with a valid excuse (illness, etc.), see me and I will excuse you from the exam. Each of the class period exams you take will then count 22.5% with the final 45% and recitation 10%. If you miss two hour exams with a valid excuse, the final will count 60%, the remaining exam 30% and recitation 10%.
- (5) <u>Final Exam</u>: Final exams are given in December just before Christmas break. If you have an "official" conflict (defined on page 110 of the Columbia College 2003-04 Bulletin and on page 198 of the Fu-SEAS 2003-04 Bulletin as 3 exams on the same day or two exams at the same time) that prevents you from taking the exam at the regularly scheduled time, you should: ....See me immediately.

If you miss or expect to miss the exam for any other reason, you must request the grade of INC from the Dean's Office in your school (see page 111 of the Columbia College Bulletin for 2003-04 and page 198 of the Fu-SEAS Bulletin for 2003-04). I am powerless to grant INC's! The Dean's Office will be very sympathetic to illness or family emergencies as reasons to miss final exams. (Deans are not sympathetic to excuses based on airline schedules, so be careful when you make your plans to return home for intercession! The final exam in this course will **most likely** be held on Tuesday afternoon, December 16, 2003; however, I cannot guarantee this date as the Registrar sets the exam schedule around November 1.) Make-up final exams for INC grades (once granted by the Dean's Office) have to be arranged with me for early in the spring term (late January or early February).

- (6) Office Hours: My office is located in 508 Havemeyer Extension and my mailbox is 3109 in Havemeyer. My University telephone number is 854 4162; FAX: 854 8336; Email: flynn@chem.columbia.edu. Office hours will be held immediately after class on TTh (2:30-3:00) or by appointment. If you would like to see me, please stop after lecture or telephone and make an appointment. I will be happy to talk with you.
- (7) <u>Homework:</u> Problem assignments are suggested about every 10-20 days. Solutions to the odd numbered problems are in the Student Solutions Manual (see item 9, "Texts", below). Solutions to even numbered problems from the text or other assigned problems not in the text will be posted on the course web site. Except for part of the first set of homework , which you must hand in, problem sets are <u>not</u> collected. Occasionally homework problems will appear verbatim on exams.
- (8) <u>Regrading of Exams</u>: Sometimes exam graders make mistakes. Regrading of hour exams will be allowed with the following rules:

(a) Clerical, addition, or numerical errors will be rechecked upon request without penalty.

(b) Other requests for regrading of any kind will result in regrading of the entire exam.(c) If you decide to hand in your exam for regrading be sure NOT to ALTER it in ANY WAY. We Xerox a large fraction of the exams before returning them to you in order to discourage that small

fraction of students who might be tempted to "bend the rules in their favor."

(9) <u>Texts:</u> The required text for this course is <u>Principles of Modern Chemistry</u> 5th Ed., by Oxtoby, Gillis, and Nachtrieb, Brooks/Cole Thomson, 2002. An old text which you may also find useful is <u>University Chemistry</u>, 3rd Ed. by B. H. Mahan, Addison Wesley, 1969. We used this text for about 20 years and generally liked it very much. However, the book is rather old in both style and content. The main advantage of this text is its rigor and a really superb set of challenging problems.

During the last quarter of the term, when I discuss chemical bonding, you may find useful a little book <u>Solids and Surfaces: A Chemist's View of Bonding in Extended Structures</u> by Roald Hoffmann, VCH Publishers, 1988, New York. Roald graduated from Columbia back in the late 50's and had the good taste and intelligence to win the Nobel Prize in Chemistry! Another useful book on chemical bonding is the paperback text <u>Chemical Bonds: An Introduction to Atomic and Molecular Structure</u> by Harry B. Gray, Benjamin/Cummings Publishing Co. 1973. Harry taught chemistry at Columbia during the halcyon days of the early and mid 60's.

In addition I would strongly urge you to purchase the <u>Student Solutions Manual</u> (to <u>Principles of</u> <u>Modern Chemistry</u>, 5<sup>th</sup> Edition) by Wade A. Freeman, Brooks/Cole Thomson, 2002. This book contains detailed solutions to odd numbered problems in <u>Principles of Modern Chemistry</u>.

I would urge you not to buy any texts other than <u>Principles of Modern Chemistry</u> and the <u>Student</u> <u>Solutions Manual</u>. Use other books to clarify points not clear in <u>Principles of Modern Chemistry</u>, or to expand on material that interests you. All these books are on reserve in the Chemistry Library, 4th Floor Chandler Hall.

(10) <u>Reading Assignments:</u> I will give two types of reading references. The first is simply a reference to material that I will be covering in lecture. You should read this ONLY if you did not understand the lecture. The second type of reading will be on material <u>not</u> covered in the lecture, but for which you are <u>responsible</u>. You should read and understand this material well enough to do the homework problems assigned to cover this material. Generally, such assignments cover work which is very important but should have been well studied in high school and thus represents a review for you

(11) <u>Suggested Method of Study</u>: Many of you have excellent study habits; others do not. My own formula for success in science courses is a simple one:

(a) Come to class (I will try to make it interesting).

(b) Take good notes (You will need to be awake to do this).

(c) Read the text only when you did not understand the lecture or when you need help doing the homework problems.

(d) DO THE ASSIGNED HOMEWORK PROBLEMS.

(e) Do the homework problems again shortly before the exam but <u>use no props</u> such as the text, the solutions, or your own notes to help. <u>At least</u> 2-3 days should elapse between (d) and (e).

(f) Understand and memorize all class derivations.

(g) Take the exams and be as relaxed as possible when you do so. Do not stay up all night studying.

(h) If at all possible, try to "get into" the material (i.e., ENJOY!).

Good Luck George Flynn