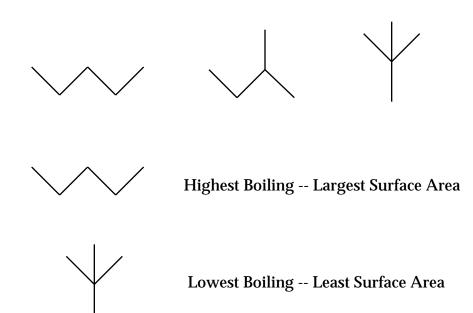
Test 1 Answers 1998

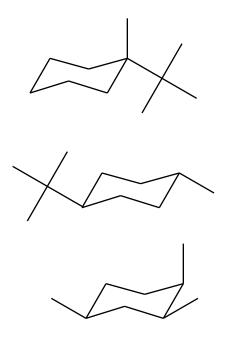
Question 1

Question 2

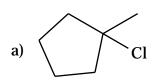
Question 3

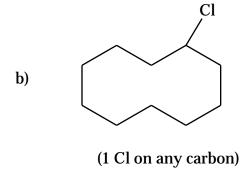


Question 4

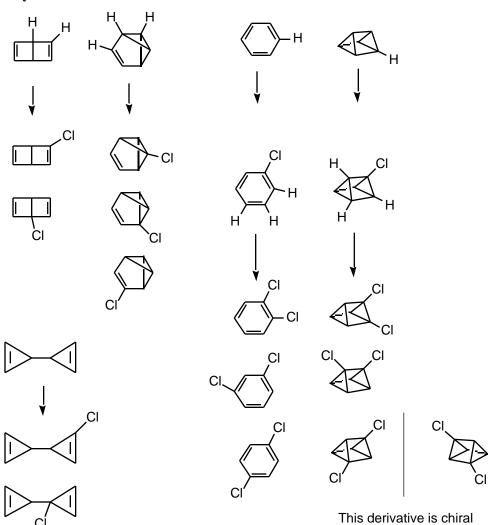


Question 5





Question 6



From the isomer count, it can be seen that only benzene and prismane yield one mono-chloro structure. This would differentiate benzene and prismane from benzvalene, Dewar benzene, and the bis-cyclopropene. Prismane and benzene also both yield three di-chloro benzenes. However, one of the prismane di-chloro isomers is chiral and yields enatiomers.

For full credit - you only had to recognize that the isomer count is solved by seeing how many different isomers could be produced.