

A long term study of the effect of elevated CO₂ on marine calcification using the Biosphere 2 coral reef mesocosm as a model system

Chris Langdon

Lamont-Doherty Earth Obs. of Columbia Uni.

Palisades, NY 10964

langdon@ldeo.columbia.edu

Outline

◆ Background

- CO₂ emissions (270 Gt C since 1751) have changed the chemistry of the atmosphere and are beginning to change the chemistry of the surface ocean.
- These changes could have a direct effect on the physiology of marine organisms.
 - ↑CO₂ may increase rates of photosynthesis in regions where nutrients are not limiting
 - ↓CO₃²⁻ may decrease rates of calcification

Outline

- If these changes were big enough the balance between photosynthesis and calcification could be upset.
 - Coral reefs could be the first ecosystems to feel the effects
 - ◆ less able to keep up with sealevel rise
 - ◆ less able to compete for space with faster growing algae
 - ◆ more susceptible to boring organisms and storm damage
 - Negative effects could extend to seagrass and mangrove ecosystems that are commonly associated with coral reefs
 - ◆ loss of important fisheries
 - ◆ loss of income associated with tourism mostly falling on poor third world countries

Outline

- If the effects extend to other and perhaps all calcifying organisms
 - a CO₂ dependent reduction in the rate of global calcification would constitute a negative feedback on the rate of increase of atmospheric CO₂ of ~0.3 Gt C y⁻¹
 - this wouldn't make a dent in the ~3 Gt C y⁻¹ that is currently accumulating in the atmosphere but it could have been important in the past in stabilizing atmospheric CO₂
 - during glacial-interglacial cycles the strength of marine calcification could have varied by 20% or 0.1-0.2 Gt C y⁻¹

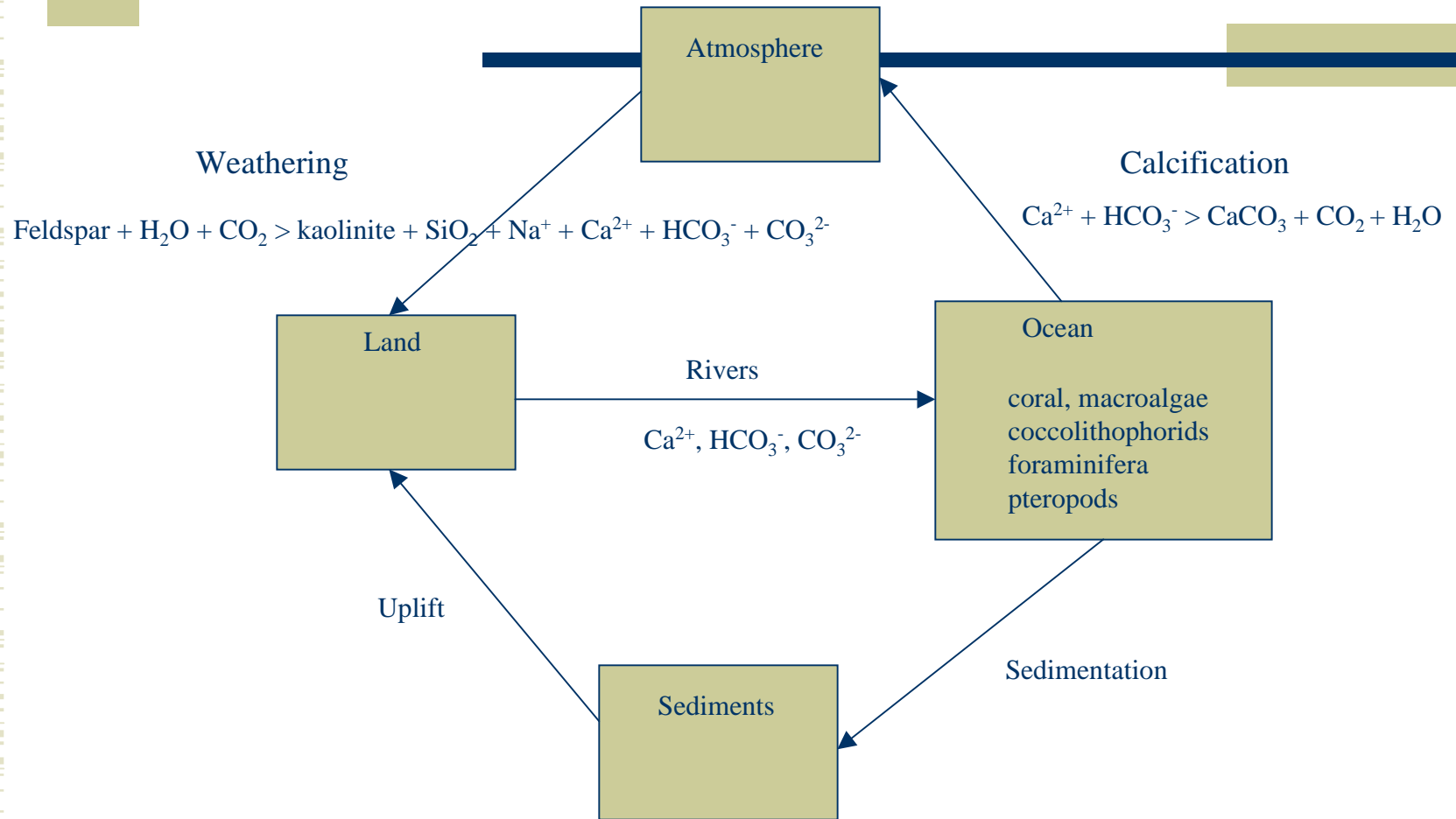
Outline

- ◆ How do we get some answers?
 - Historical approach
 - Detailed records of coral growth over glacial-interglacial cycles not available
 - Expected response over the last 200 years is only 10%, probably too small to separate from other environmental factors
 - We can start monitoring coral growth and Ω now but by the time we have definitive results it may be too late.

Outline

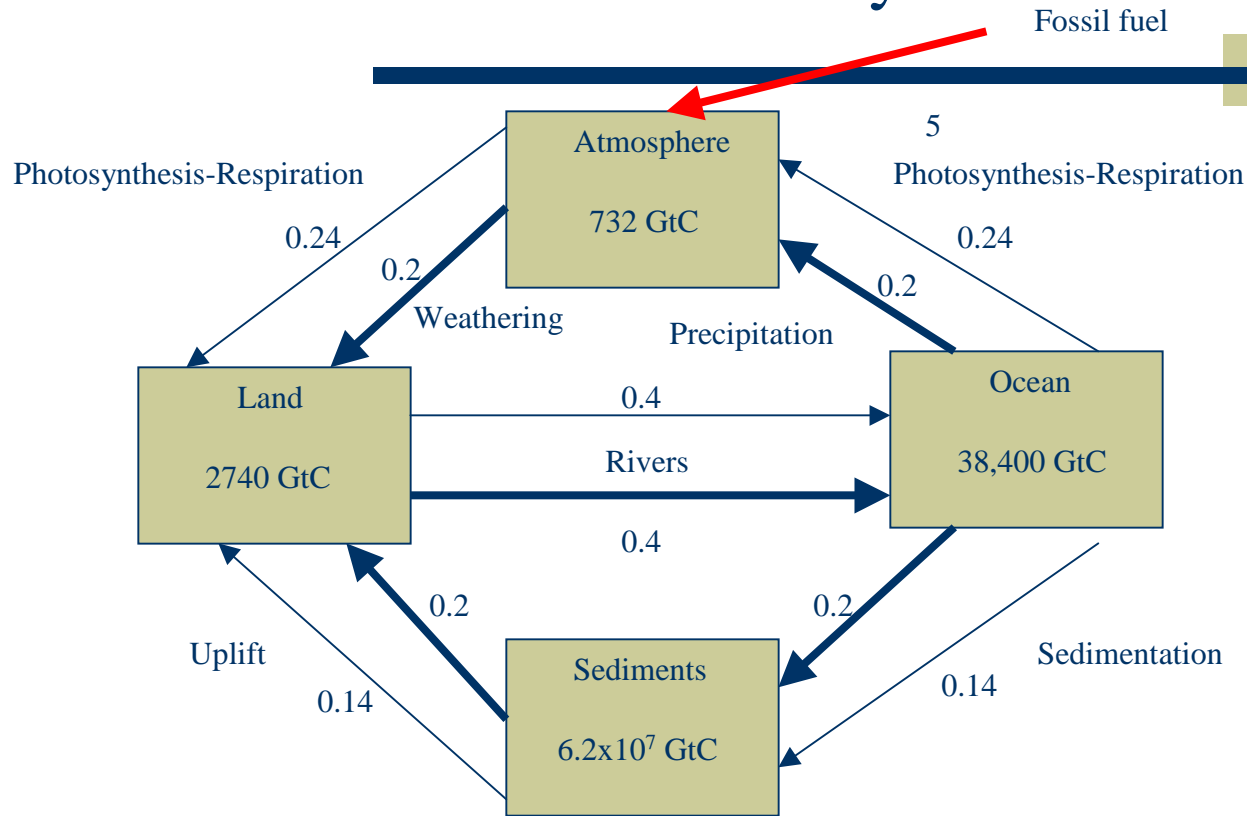
- Experimental approach
 - Manipulate carbonate chemistry to simulate future conditions and observe the effect on calcification and photosynthesis on different time scales
 - Need to choose a model system
 - ◆ Shallow tropical environments contribute 40% of global carbonate production. Coral reefs are the archetype community for these environments and may be especially sensitive to rising CO₂ as mentioned earlier.
 - ◆ B2 offers a coral reef community where long term, controlled experiments are possible

Marine carbonate cycle



Global carbon cycle

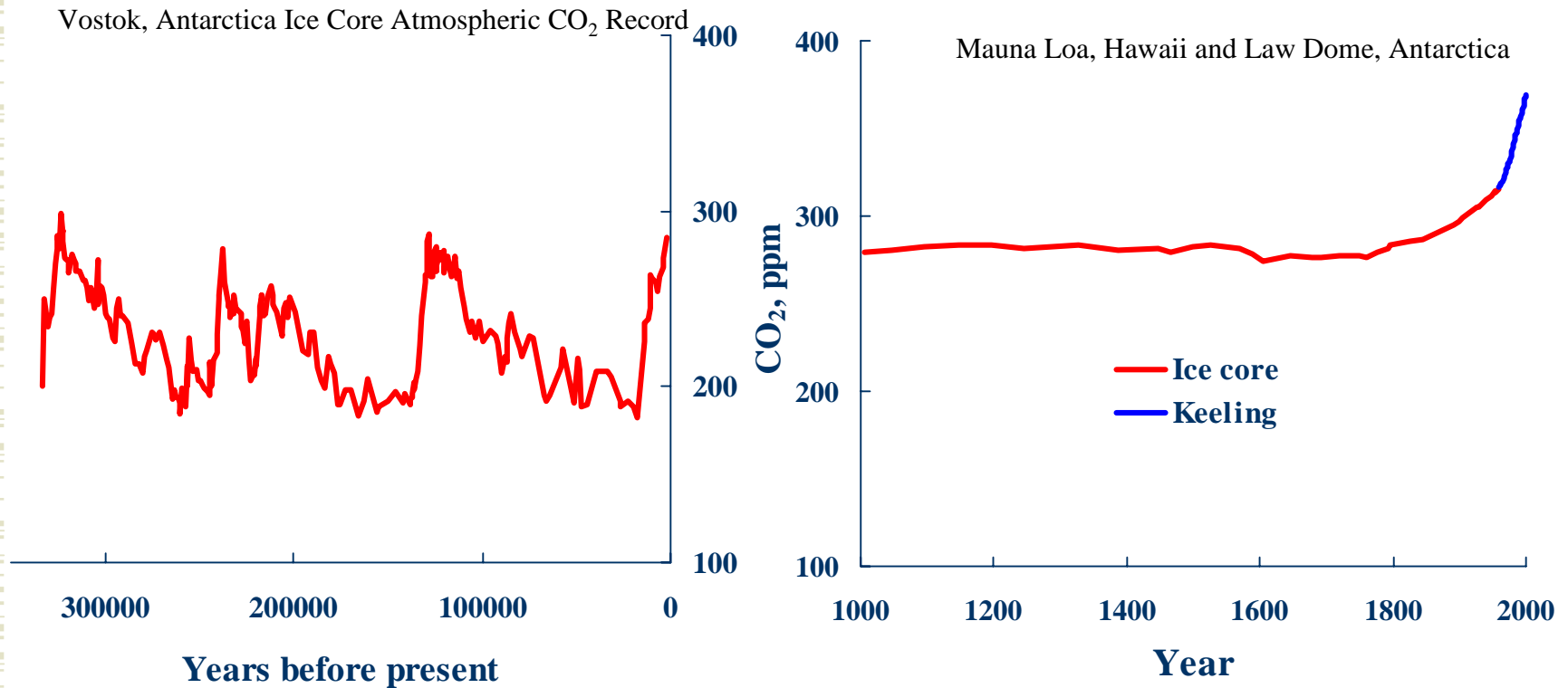
fluxes in GtC y⁻¹



 Inorganic carbon
 Organic carbon

Wollast and Mackenzie, 1989

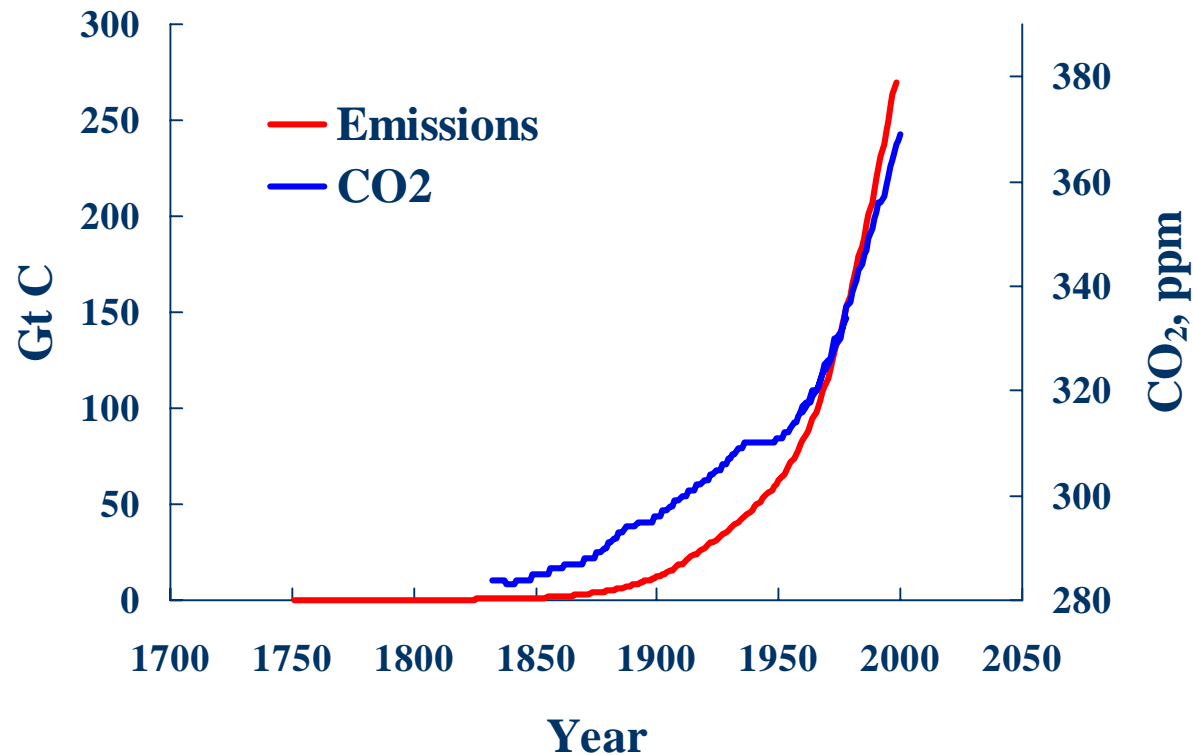
Trends in atmospheric CO₂



Barnola et al. (1999)

Etheridge et al. (1998) and Keeling Whorf (2001)

Global Emission of CO₂



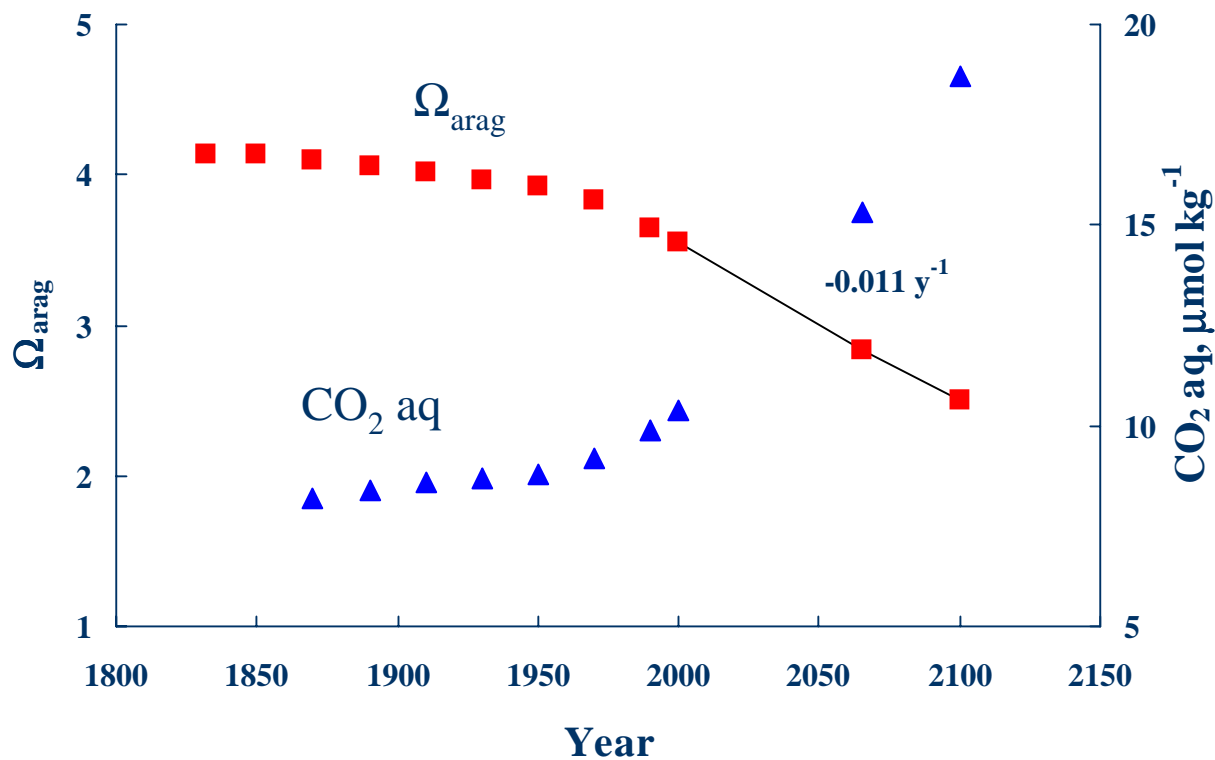
Marland, Boden and Andres (2001)

Link between atmospheric CO₂ and the rate of calcification

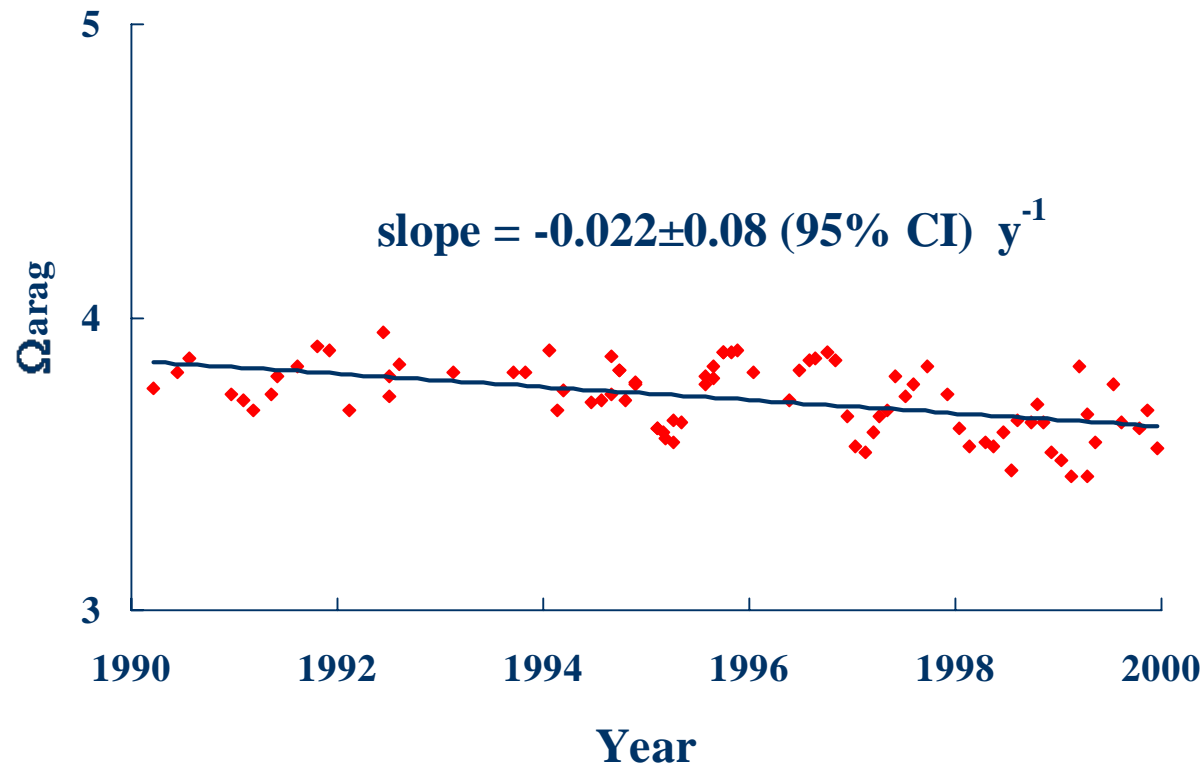
- ◆ $\text{CO}_2 + \text{CO}_3^{2-} + \text{H}_2\text{O} \leftrightarrow 2\text{HCO}_3^-$
- ◆ $\Omega_{\text{arag}} = [\text{Ca}^{2+}][\text{CO}_3^{2-}]/K_{\text{sp}}$
- ◆ $R = k(\Omega - 1)^n$

Calculated changes seawater carbonate chemistry

(assuming IS92a business as usual scenario, $S=35$, $TA=2300$)



Observations at the Hawaii Ocean Times Series Station



Biosphere 2 coral reef biome



Corals in B2 ocean



Montipora



Siderastrea

Porites



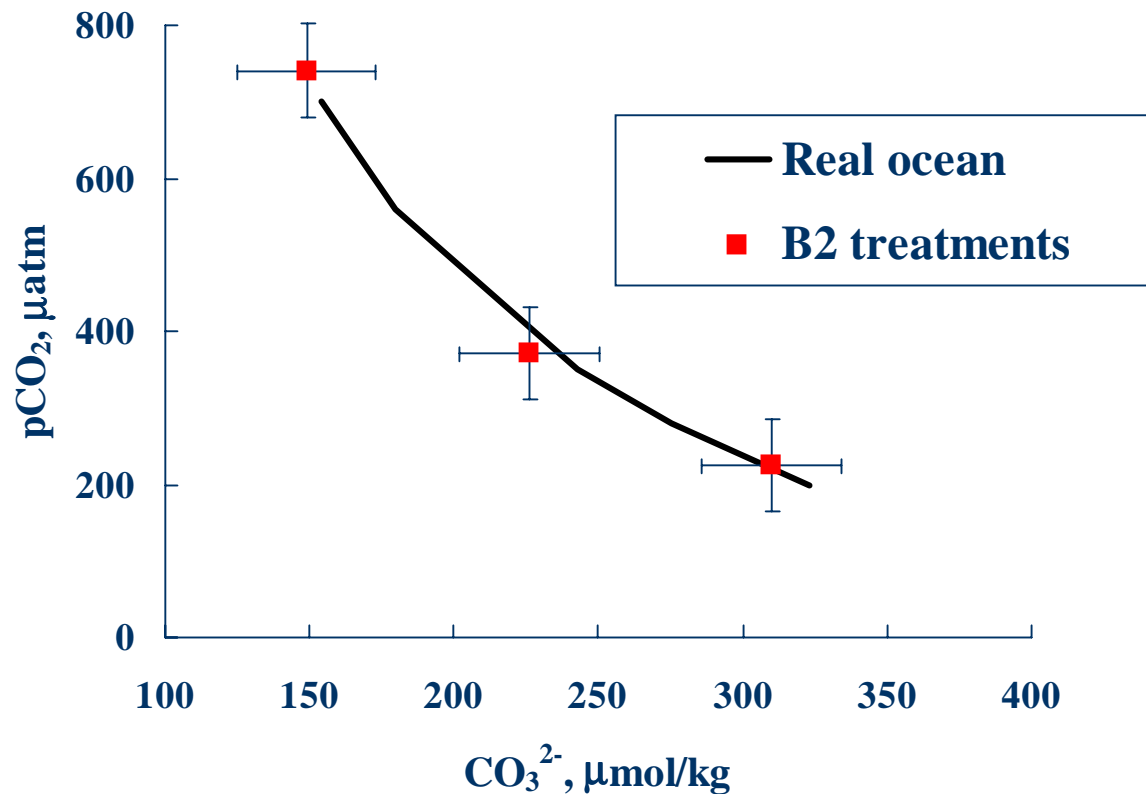
Experimental design

- ◆ Simulate real world changes by varying $p\text{CO}_2$ between 200, 350 and 700 μatm at regular 4 monthly intervals and observe the effect on community net primary production and calcification.
- ◆ Manipulate $p\text{CO}_2$ by adjusting TA with additions of HCl and NaOH while holding TCO_2 constant by additions of NaHCO_3 and Na_2CO_3 .

Chemical treatments

	n weeks	pCO ₂ μatm	HCO ₃ ⁻ μmol kg ⁻¹	CO ₃ ²⁻ μmol kg ⁻¹	pH sws
LGM	29	225±29	1546 ±70	310±28	8.23±0.04
PD	47	371±55	1696±76	226±25	8.05±0.06
FUT	43	741±99	1954±45	149±18	7.81±0.06

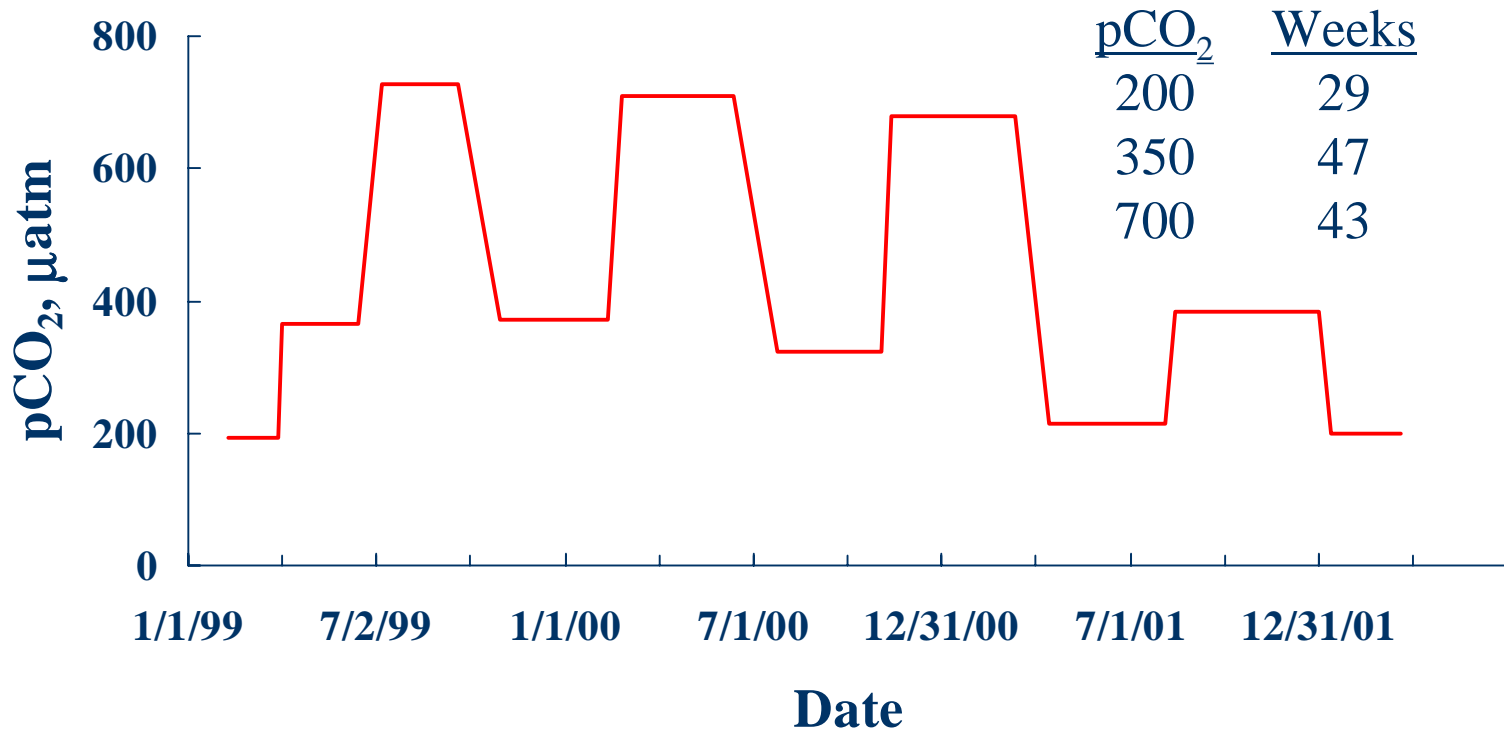
Comparison of experimental manipulations and real world changes



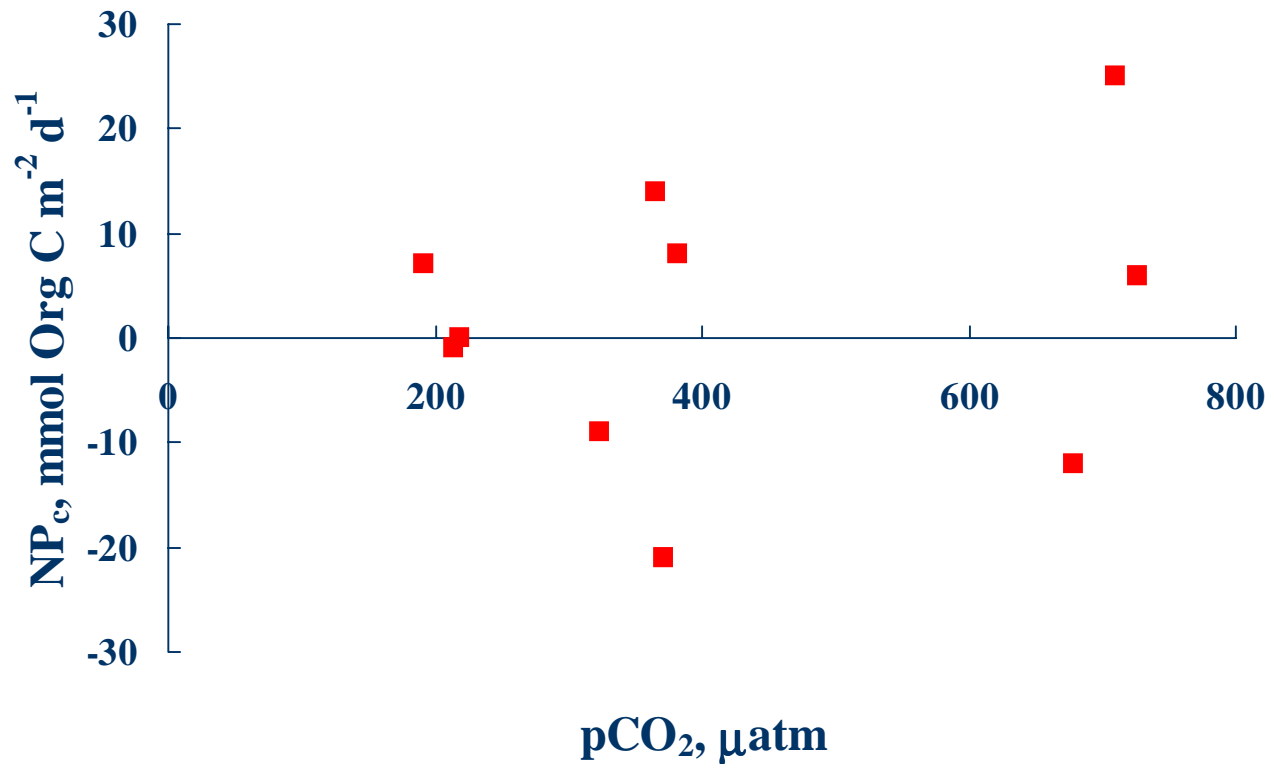
Calculation of Calcification and Net production

- ◆ $G \text{ (mmol CaCO}_3 \text{ m}^{-3} \text{ d}^{-1}) = -0.5(3.5)\Delta TA/\Delta t$
- ◆ $NP_c \text{ (mmol Org C m}^{-3} \text{ d}^{-1}) = 3.5(\Delta TCO_2/\Delta t - 0.5\Delta TA/\Delta t) + (1.8 \text{ m d}^{-1})(0.027 \text{ mmol m}^{-3} \mu\text{atm}^{-1})(pCO_{2,w} - pCO_{2,a})$

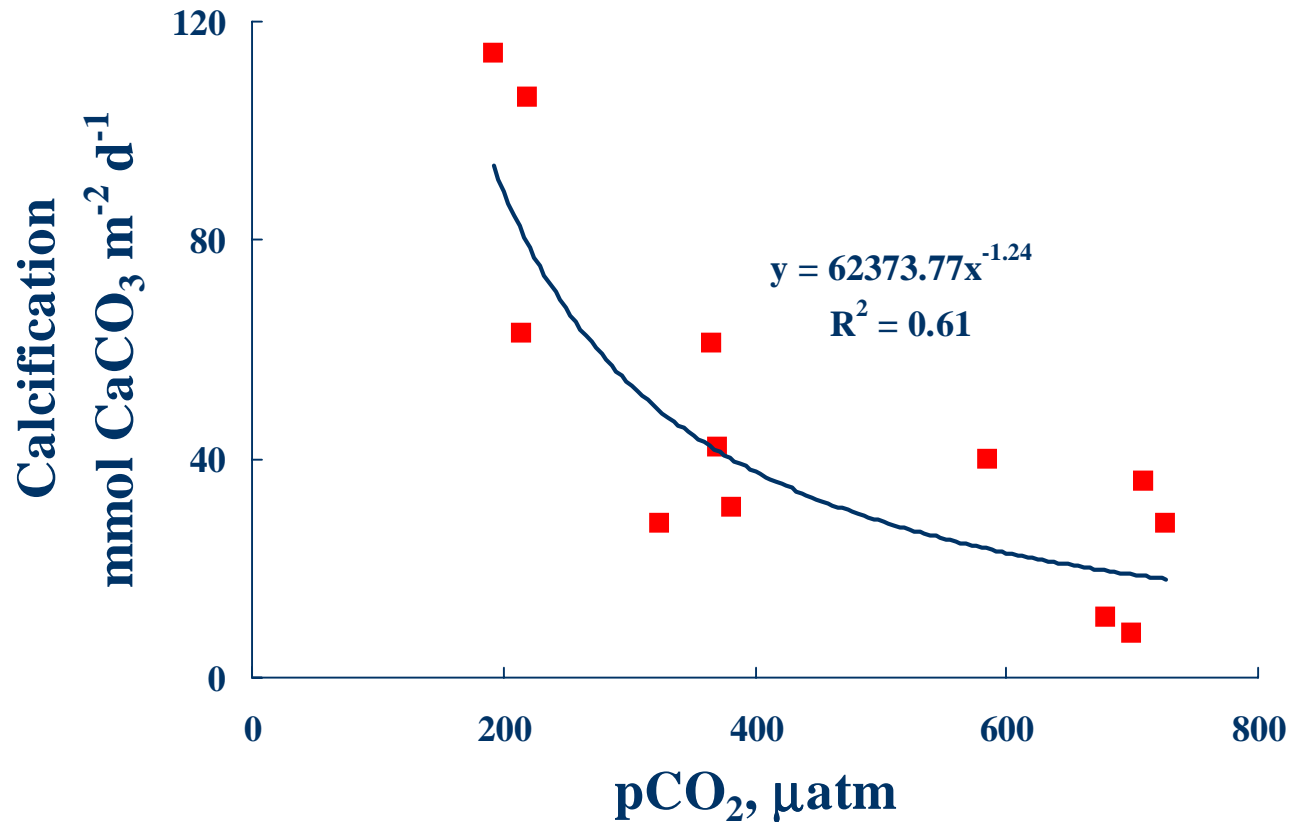
Time line of CO₂ treatments



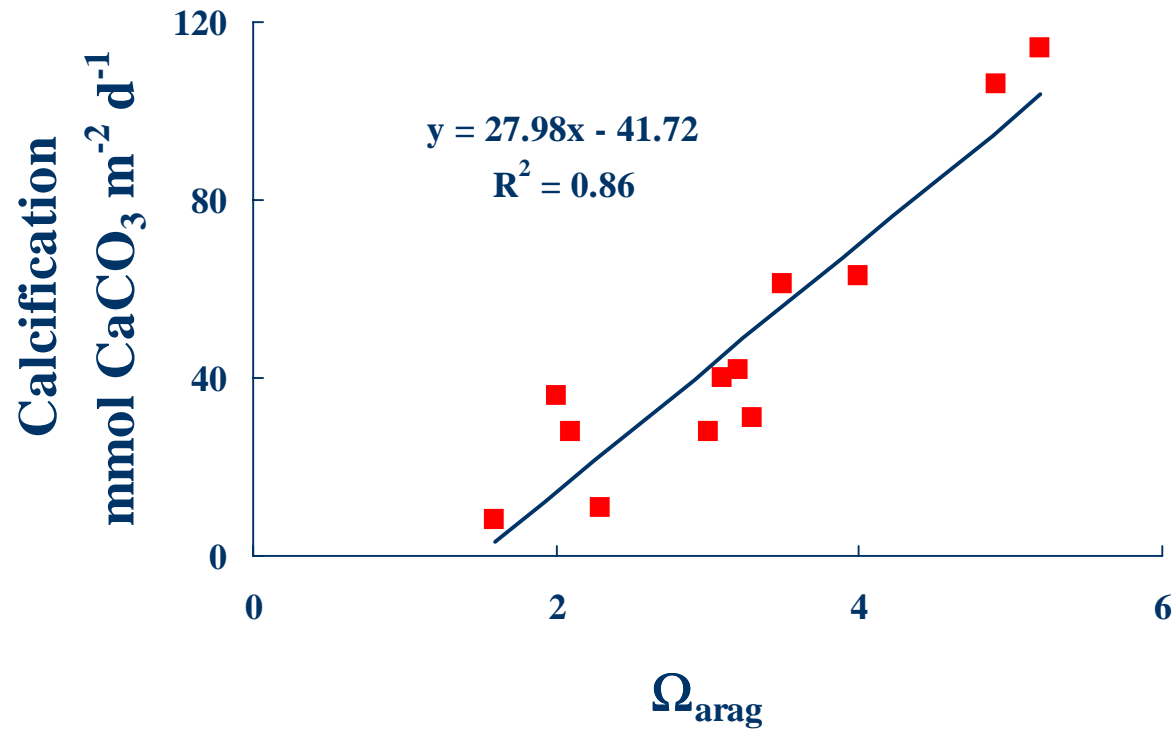
Effect of CO₂ on net community production



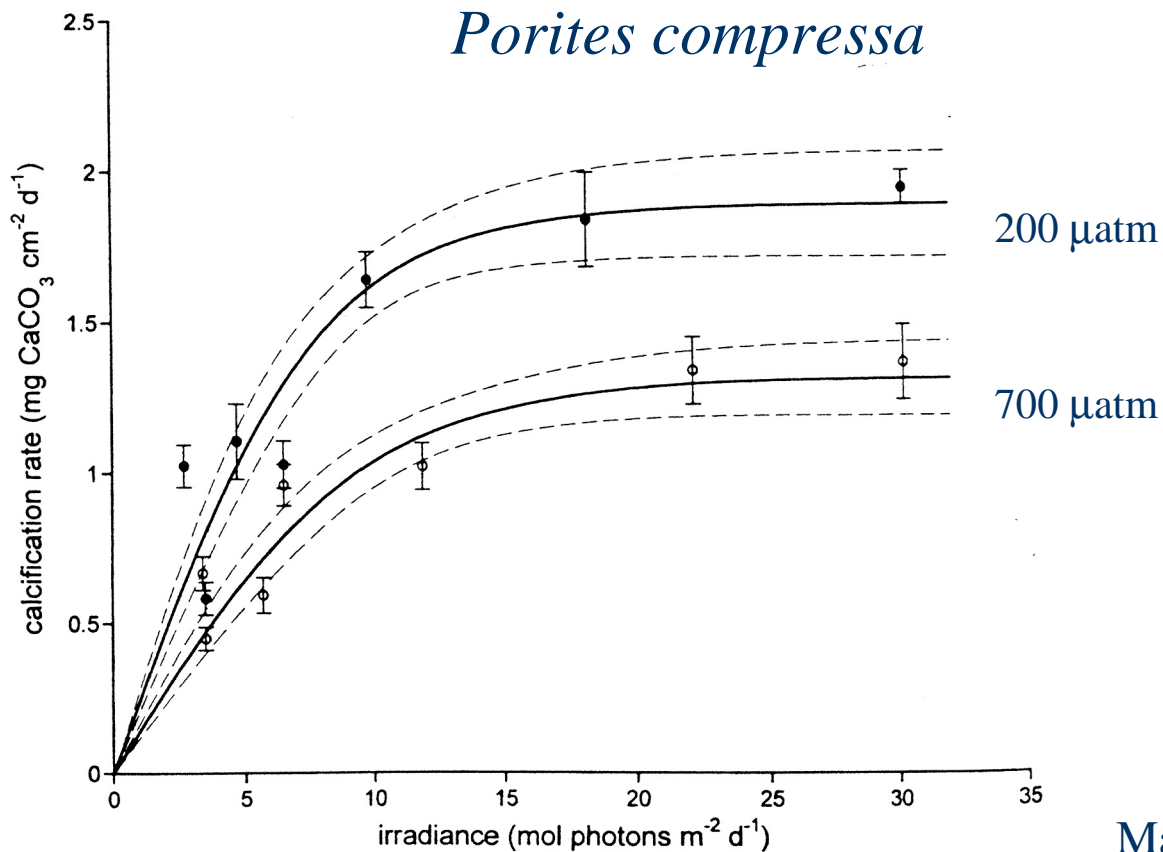
Effect of CO₂ on community calcification



Saturation state controls calcification

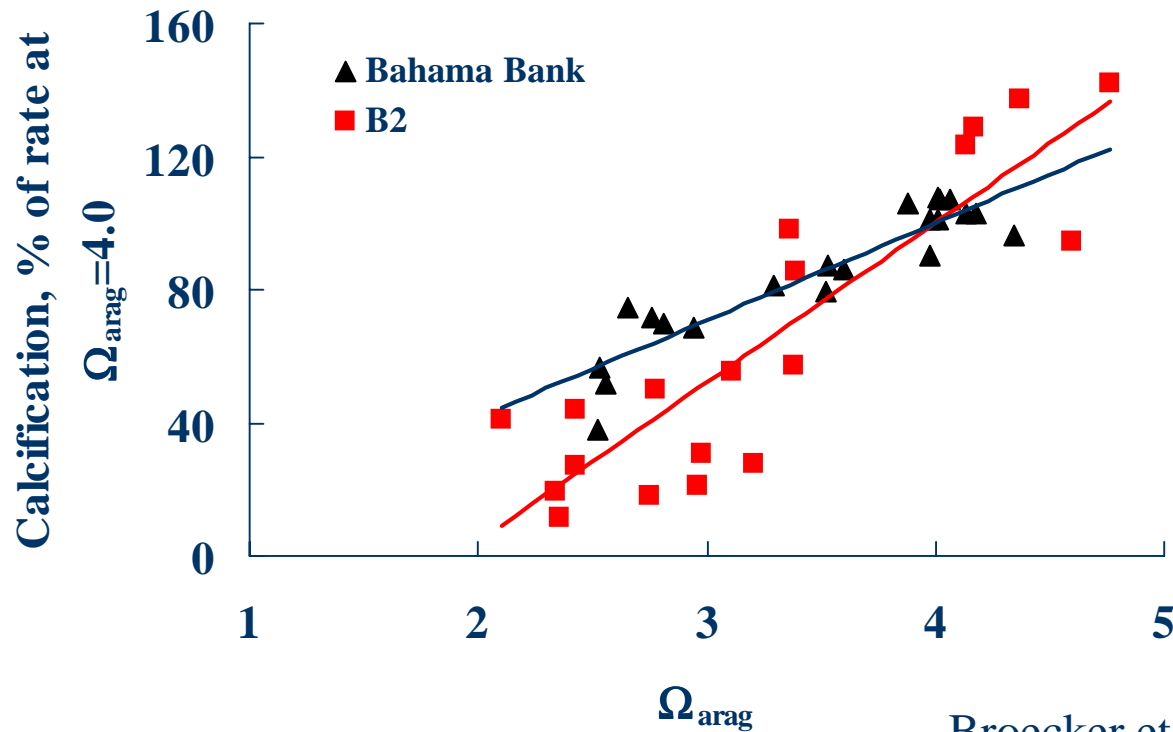


Phase 3- Species specific responses



Marubini et al., 2001

Comparison of B2 with Great Bahama Bank



Broecker et al., 2001

Effect of a doubling in CO₂ (350-700) on calcification, (% decrease)

Calcareous macroalgae

<i>Amphiroa foliacea</i>	-36	Borowitzka, 1981
<i>Porolithon gardineri</i>	-16	Agegian, 1985
<i>Corallina pilulifera</i>	-44	Gao et al., 1993

Corals

<i>Stylophora pistillata</i>	-3	Gattuso et al., 1998
<i>Porites porites</i>	-16	Marubini & Thake, 1999
<i>Porites compressa</i>	-27	Marubini et al., 2001
<i>Acropora sp.</i>	-37	Schneider & Erez, 2000
<i>Porites/Montipora</i> prep.	-50	Langdon & Atkinson, in

Coccolithophorids

<i>Emiliana huxleyi</i>	-10	Riebesell et al., 2000
<i>Gephyrocapsa oceanica</i>	-29	“ “
Natural pop. (N. Pac.)	-38	“ “
<i>Emiliana huxleyi</i>	-17	Zondervan et al., 2001
<i>Gephyrocapsa oceanica</i>	-29	“ “

Community

Biosphere 2	-40	Langdon et al., 2000
Monaco mesocosm	-21	Leclercq et al., 2000
Bahama Bank	-30	Broecker & Takahashi, 1966

Conclusions

- ◆ Photosynthesis and calcification are not tightly coupled with respect to their response to rising CO_2 .
 - $\uparrow\text{CO}_2$ NP_c unchanged $\downarrow\text{Calcif.}$
- ◆ There is a nonlinear relationship between calcification and pCO_2
 - for 200 to 280 μatm $\text{pCO}_2 \uparrow$ Calcif. \downarrow 34%
 - for 350 to 700 μatm $\text{pCO}_2 \uparrow$ Calcif. \downarrow 58%

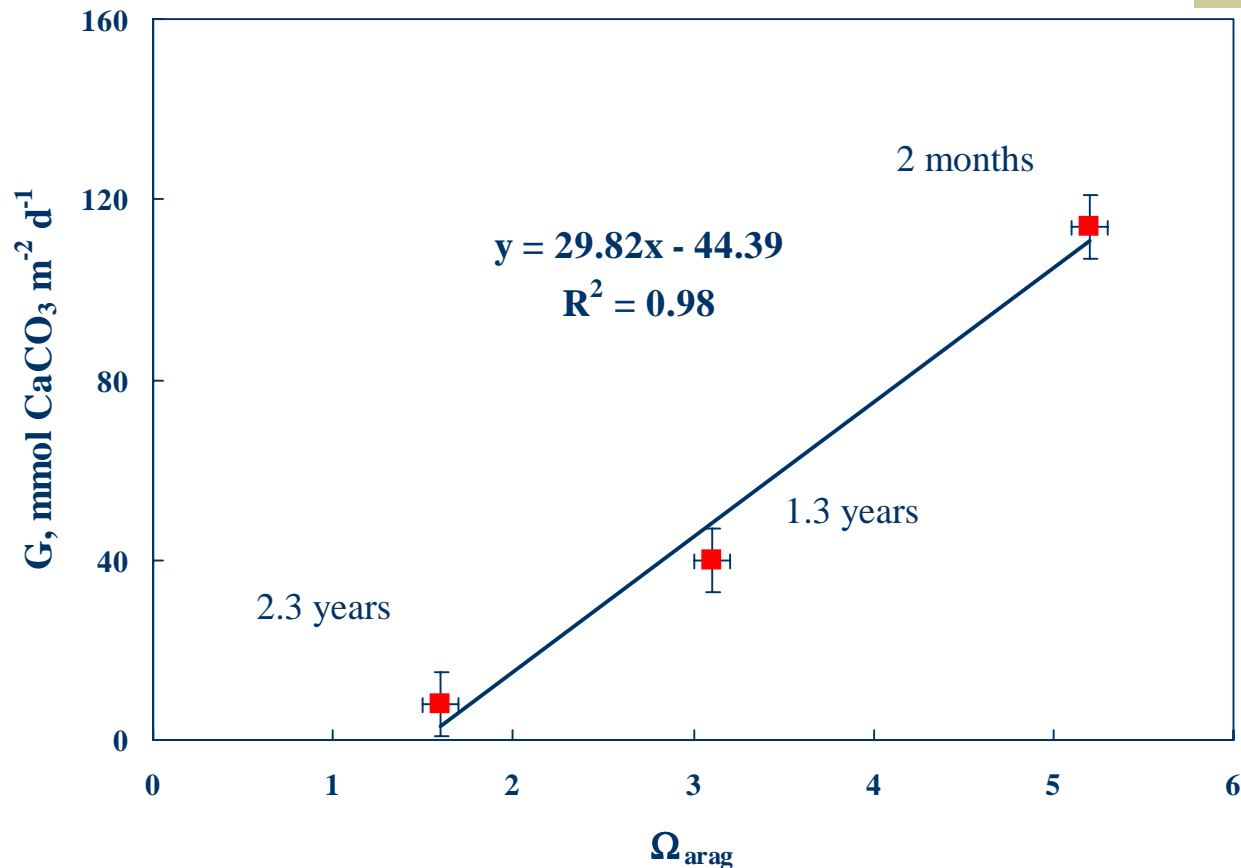
Conclusions

- ◆ Data are consistent with the hypothesis that saturation state (Ω) controls the calcification of the B2 coral reef system, several species of coral, calcareous red algae, coccolithophorids and a natural community dominated by green algae.
- ◆ Consequences of reduced calcification to corals and coral reefs
 - reduced ability to compete for space and light
 - reduced ability to keep up with sealevel rise
 - increased susceptibility to erosion and damage by fish, boring organisms and storms.

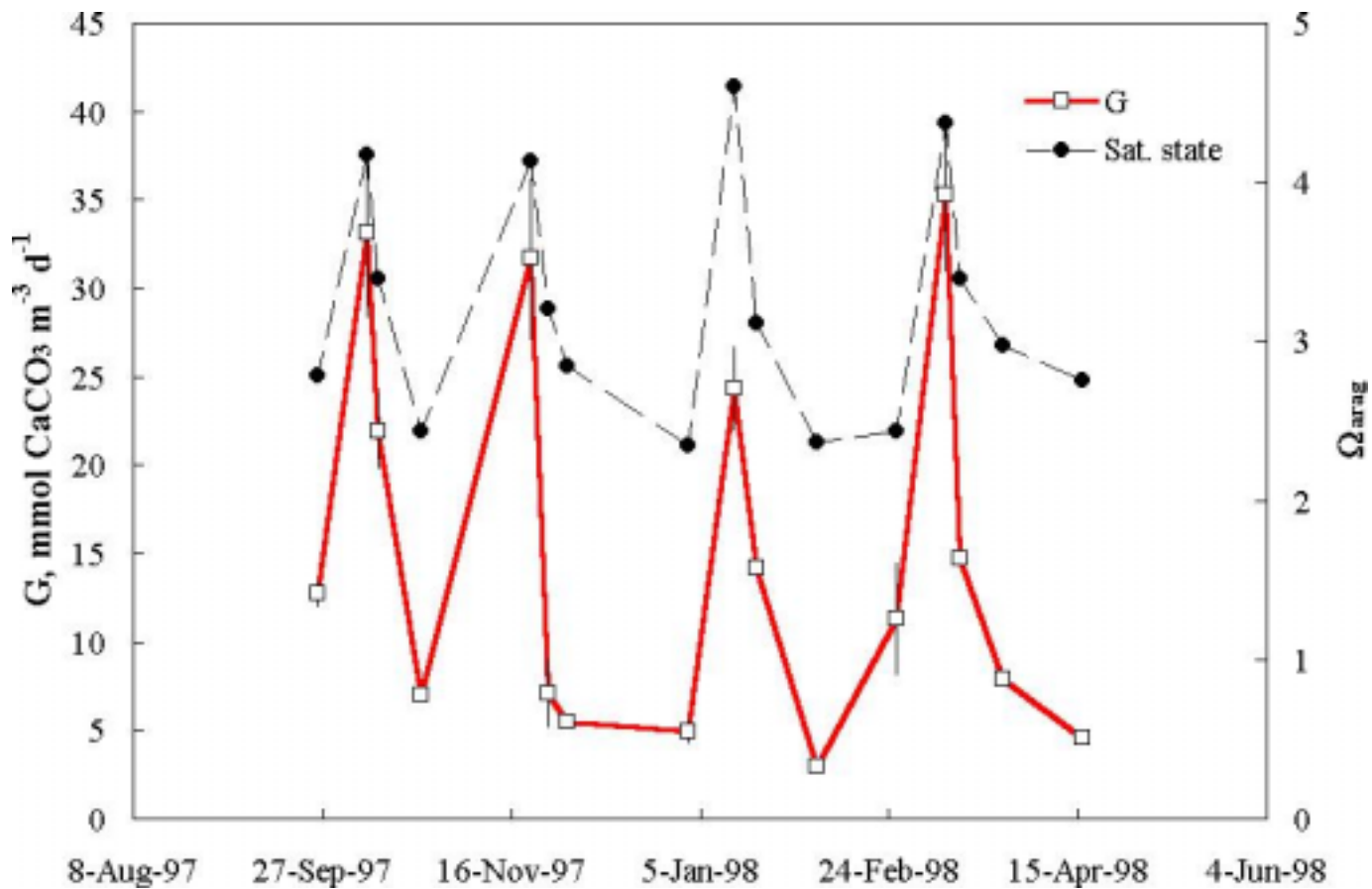
Conclusions

- ◆ From a geochemical point of view the reduction in the ratio of calcification to net photosynthesis as CO₂ rises provides a negative feedback on atmospheric CO₂ levels.
- ◆ Such a feedback mechanism was first suggested based on laboratory culture experiments here we show that the mechanism can also be demonstrated for a very different community of organisms and at a much larger spatial and longer temporal scale.

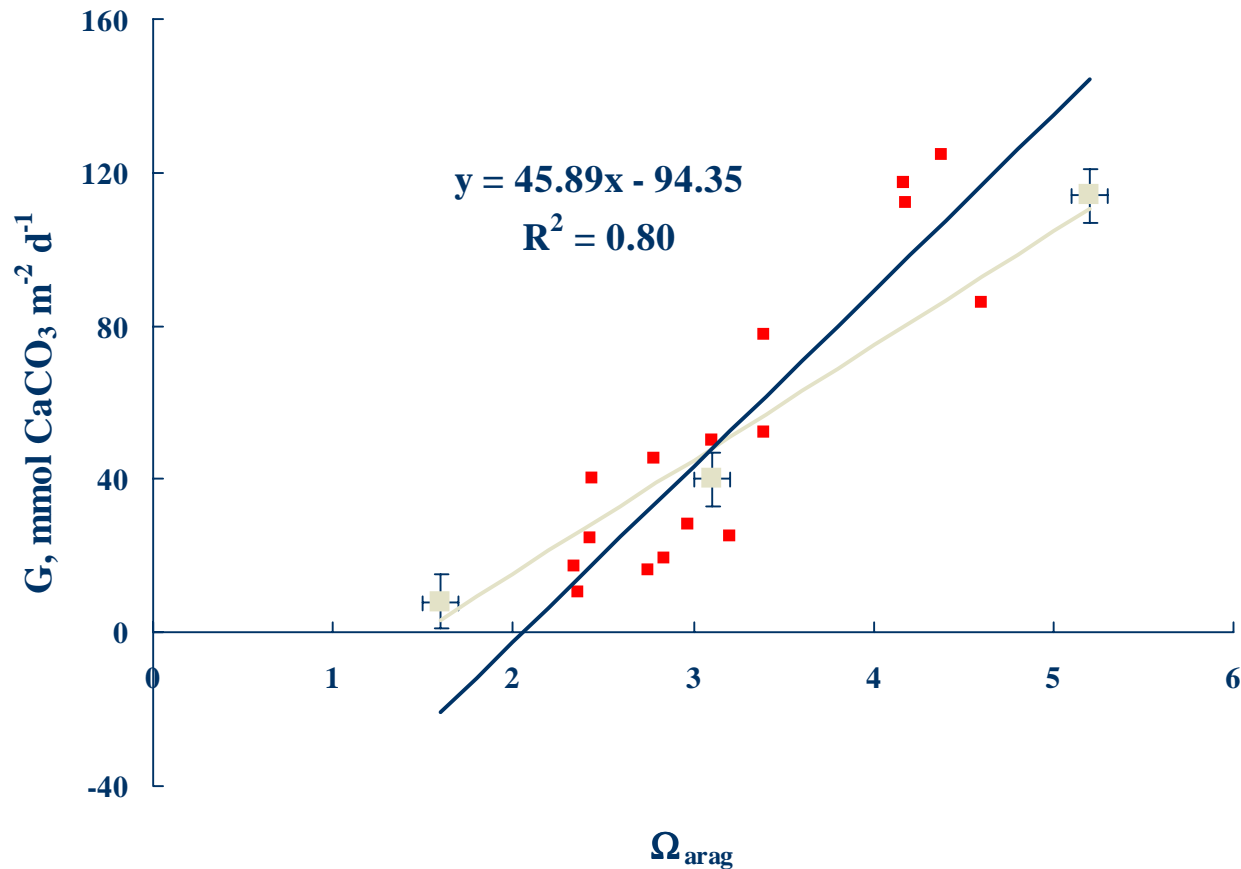
Phase 1 – long-term response



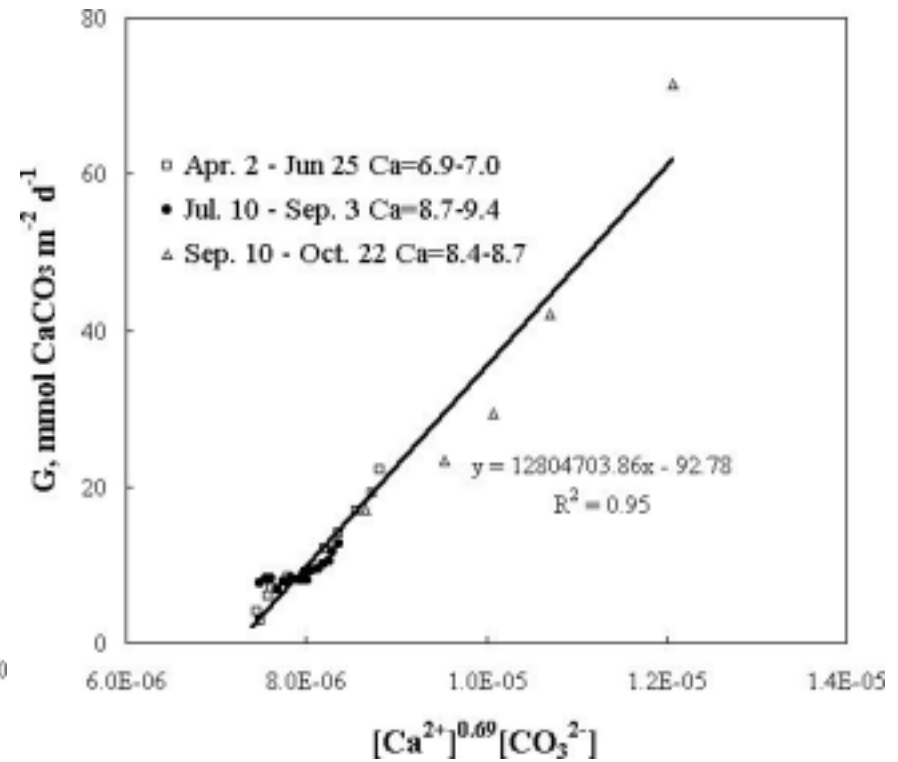
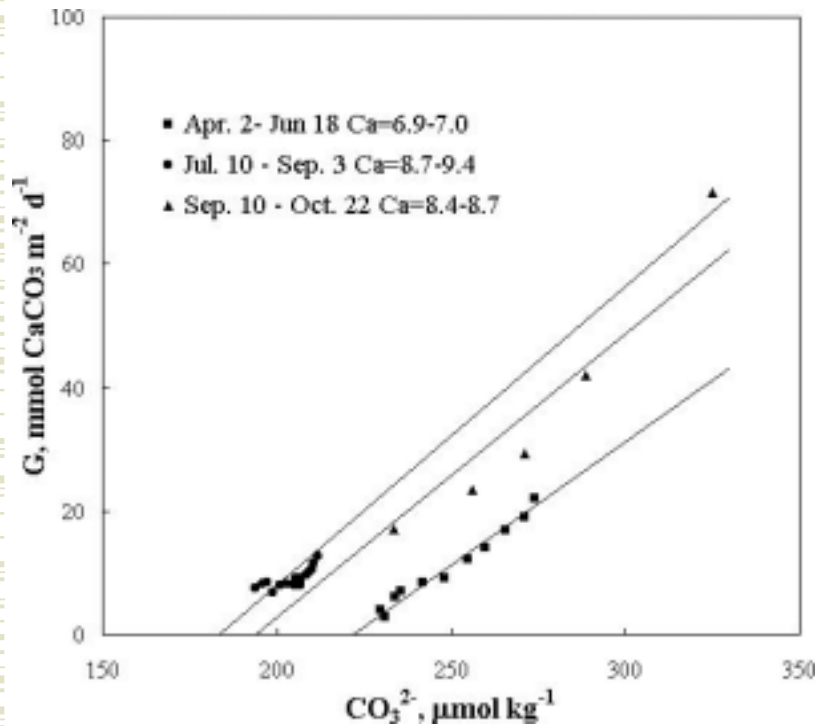
Phase 2 – rapid flip flopping of Ω_{arag}



Phase 2 – short term response

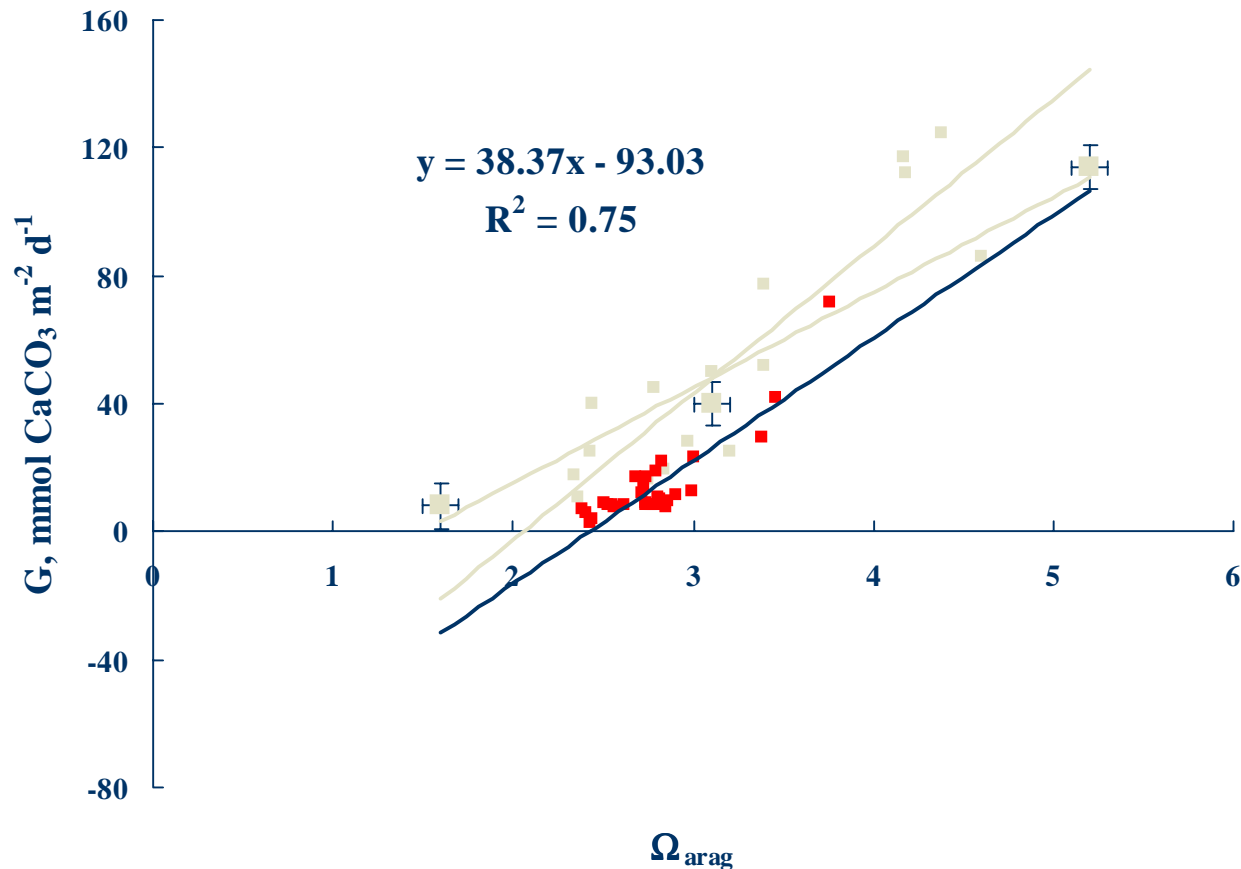


Phase 2 – test of saturation state hypothesis



Langdon et al., 2000

Saturation state dependence – all time scales



Conclusions

- ◆ Studies performed in different ways all show that calcification of corals and coralline algae will decline due to rising CO₂.
- ◆ Production of framework and fill carbonate on coral reefs will decline to 50-97% (avg. 70%) of 1880 rates by the year 2065.

Implications

- ◆ Corals may produce weaker more easily damaged skeletons.
- ◆ Linear growth of corals may decrease making them less able to keep up with rising sealevels.
- ◆ Due to latitudinal gradient in Ω high latitude coral reefs should be the first to be affected.

Future work

- ◆ Need to determine effect of reduced calcification on skeletal density and linear growth rate of corals.
- ◆ Need to start monitoring calcification or carbonate accumulation on a few natural reefs. Many things might cause a reduction in calcification but if a latitudinal pattern emerges with high latitudes sites showing the greatest reduction the CO₂ hypothesis would be supported.