

Problem Set #20

73. Find the effective inter-nucleus potential caused by the orbiting electron for the ground state of the H_2^+ molecule, using the Born-Oppenheimer approximation. You may follow the treatment given in Baym, using an electron wave function constructed from a superposition of two hydrogenic wavefunctions, each centered on one of the protons.
74. Two identical spin-1/2 particles of mass m are confined by an infinite potential to move in a 3-dimensional box of side L . They interact with a spin-independent potential $V(\vec{r}, \vec{r}') = \lambda \delta(\vec{r} - \vec{r}')$.
- (a) Find the three lowest energies of the allowed two-particle states when $\lambda = 0$.
 - (b) What are the corresponding eigenstates?
 - (c) Find the corrections to the energies of these eigenstates to first order in λ , being careful to deal correctly with degeneracy.
75. Imagine that the electrons in an atom with nuclear charge Z are actually bosons with spin zero.
- (a) Find the ground state and ground state energy for the resulting atom, neglecting the electrostatic interaction between the electrons.
 - (b) Calculate the shift in the ground state energy determined in part (a) to first order in the potential describing the electrostatic repulsion between the orbiting electrons.
 - (c) If these bosonic electrons have spin 1, what is the degeneracy of ground state neglecting any magnetic effects associated with the particle spins?
 - (d) Assume these identical, spin-1 particles have magnetic moment $\vec{\mu} = \gamma \vec{J}$ and that the atom is placed in a uniform magnetic field $\vec{B}$. Find the new ground state, its energy and degeneracy, treating the magnetic field as weak. Again neglect any spin-spin interactions between the bosons.
 - (e) For the situation studied in part (d) find the first and second excited states, their energies and degeneracies.